GROWTH OF AlN CRYSTAL USING REACTIVE SPUTTERING METHOD WITH VARIATION OF NITROGEN GAS PRESSURE AND SUBSTRATE TEMPERATURE

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ABSTRAK


Kata kunci: Semikonduktor, aluminium nitrida (AlN), SiO₂, Reactive Sputtering, Transversal Optik

ABSTRACT

GROWTH OF AlN CRYSTAL USING REACTIVE SPUTTERING METHOD WITH VARIATION OF NITROGEN GAS PRESSURE AND SUBSTRATE TEMPERATURE. Today’s semi-conductor technology continues to develop, both in terms of material discovery, manufacturing technique, as well as implementation and the development. In this research, the researcher will utilize the AlN synthesis with a reactive sputtering method on the substrate SiO₂ that has done with variation substrate temperature of (200, 250, 300°C), and the nitrogen gas pressure of 10, 15 and 20 mBar. This study successfully synthesize nanomaterial AlN with a hexagonal crystal system and shows the wurztite phase is formed. While morphology AlN formed composed of particles - spherical shaped particles. From this research it is known that raising the temperature of the substrate produces more uniform morphology and homogeneous, the surface roughness is formed from 15.4 to 84 nm and obtained the crystal size is 64-87 nm. While the addition of nitrogen gas pressure causes the better crystalallinity. The research obtained AlN with Optical Transversal wave type at a wavenumber of 600-670 cm⁻¹. The growth of AlN crystal growth of AlN is best obtained on the parameters of the substrate temperature 290 °C and nitrogen gas pressure of 20 mbar.

Key words: Semiconductors, aluminum nitride (AlN), SiO₂, Reactive Sputtering, Optical Transversal

INTRODUCTION

One of semi-conductor materials that attracts today’s attention is Aluminum nitride (AlN). AlN is an important member of the group III-V with the highest bandgap of around 6.2 eV. It is a promising advanced ceramic material with many interesting properties, such as high
electrical resistivity (>10\(^{14}\) Ω·cm), high thermal conductivity (285 W/(m·K)) \[1\]. Therefore, AlN is a potential candidate for many applications on fusion fuel nuclear, solar cells, transparent conductive coatings, gas sensors, semiconductor materials as well as electro and photoluminescence. Several methods of physical and chemical synthesis to produce AlN with a 1-dimensional nanostructures have been developed, including vapor phase method such as thermal evaporation, chemical vapor deposition, deposition of metal-organic vapor phase (MOCVD), electrodeposition and synthesis methods such solutions soles gel, solution deposition, hydrothermal synthesis, microemulsion technique, as well as direct growth the solution of alcohol-aqueous \[2\], has been made to the process of formation of AlN, for example, the growth of product AlN can occur at temperatures 1300\(^\circ\)C at a rate of 10 \(^{\circ}\)C/min, using the technique of Vapor Liquid Solid (VLS) and using a stream of nitrogen gas \[3\].

RESULT AND DISCUSSION

Fourier Transform Infra Red (FTIR) Analysis

The purpose of testing Fourier Transform Infra Red (FTIR) is to identify a compound based on functional clusters. To determine the infrared absorption AlN compound that can be used in the application light emitted diode (LED). In this study, spectra of graphs showing different characteristics. FT-IR testing is done by using Shimadzu 8400S test data obtained from the top of the chart obtained is at wavenumber range between 400- 4000cm\(^{-1}\).

![Figure 1. The results of Fourier Transform Infrared Spectroscopy (FT-IR) on the AlN substrate temperature to 250\(^\circ\)C and pressure of nitrogen (a) 10 mBar (b) 15 mBar (c) 20 mBar](image)
Figure 1 shows that, at a temperature of 250°C from the test results FT-IR peak have varied from 97-99% transmittance in the range of 600-800 cm\(^{-1}\) are almost the same as the previous variation which indicates the downward peak beam then forwarded and absorbed the higher. According to the results of testing of FTIR, AlN compounds generated in this study can be applied to light emitted diode (LED). AlN compound capable of transmitting visible light at a wavelength of 600-800 cm\(^{-1}\). Wave AlN compounds are inorganic compounds (Fei et al, 2012). From the research results can be seen that the dominant wavelengths formed at intervals of 600-800 cm\(^{-1}\). In wavenumber interval 600-670 an optical transverse force active infrared and visible AlN compound produces infrared optical properties of the active transverse to the parameter variations in temperature to 250°C in nitrogen pressure 20 mBar.

In accordance with characterization LED that can emit electromagnetic waves, the compounds can be applied in a window AlN layer of the LED. At the peak absorption region of 2000-2200 cm\(^{-1}\), there are other weak peak was observed that the vibration mode of bonds N-Al-N, it further confirms that growth over SiO2 are stoichiometric AlN substrate [5]. For the C-O bond bending measured the impurities from the air were identified during the measurement process. It also deals with the homogeneity of crystals that can improve the optical properties where the target is deposited uniformly and relatively the same, making the level of crystallinity the better, so as to boost the value of transmission [6].

**Transmission Electron Microscope (TEM) Analysis**

Testing Transmission Electron Microscope (TEM) was used to analyze the morphology, crystal structure and composition of the specimen. TEM provides higher resolution than SEM and can facilitate the analysis of the size of the nanomaterial. TEM testing in this study was conducted using a JEOL JEM-1400 and conducted at the Laboratory of Chemistry, University of Gajah Mada, Yogyakarta.

TEM imaging results in Figure 2 (a) show that AlN formed a nanomaterial with magnification scale of 100 nm and 20 nm, for parameter 300°C substrate temperature and nitrogen pressure 20 mBar on measurements obtained ± 64-87 nm size. On the results of the previous morphological observations performed with the SEM imaging shows the morphology of AlN obtained less clear. In the picture TEM imaging can we know that the round-shaped particles which tend to merge so that oval and elongated.

![TEM Image](image)

Figure 2 (b) shows the test results SAED, for analysis of the results diffraction pattern obtained one point in the ring, after the calculation of the value of d-spacing. The results of the calculations show that the value of diffraction d-spacing is 1.38163 in the notation field (200). The result is worth almost the same as the value of d-spacing of XRD test results is 1.35653 in the field (200). XRD characterization results have a peak in 3 dominant fields, represented by field (101) at 2\(\theta\)=38.758°, field (103) at 2\(\theta\)=65.443°, field (200) at 2\(\theta\)=69.40°. For the highest peak intensity result at 2\(\theta\)=69.40°, with a match JCPDF03-1144.
CONCLUSIONS

Based on the research that has been done, it can be concluded that the synthesis of compounds of aluminum nitride (AlN) has been successfully performed using reactive sputtering over SiO$_2$ substrate. Of the various substrate temperature and nitrogen pressure this study we concluded that, AlN compound of testing Fourier Transmission Infrared (FTIR) showed wavenumber 600-800 cm$^{-1}$, for the FTIR results wavenumber of 600-670 cm$^{-1}$ is the Optical Transversal. TEM results of the tests show that AlN formed a nanomaterial size 64-87 nm.

REFERENCES